

**Proposed New San Diego Mechanism**

REACTIONS	A	n	E (cal/mol)	Note
O+N2<=>NO+N	1.47E+13	0.3	75286.81	No Change
N+O2<=>NO+O	6.40E+09	1	6285.85	No Change
N+OH<=>NO+H	3.80E+13	0	0	No Change
NH+H<=>N+H2	1.00E+14	0	0	J.A. Miller et al. PECS, 15 (1989) 287 More Recent Data, was Too Low
NH+O<=>NO+H	9.20E+13	0	0	No Change
NH+OH<=>HNO+H	4.00E+13	0	0	No Change
NH+OH<=>N+H2O	5.00E+11	0.5	2000.48	No Change
NH+O2<=>HNO+O	4.60E+05	2	6500.96	No Change
NH+NO<=>N2O+H	1.80E+14	-0.351	-244	P. Glarborg et al., CNF, 115 (1998) 1 More Recent Data, was Too Low
NH+NO<=>N2+OH	2.20E+13	-0.23	0	No Change
NH2+H<=>NH+H2	4.00E+13	0	3652.01	No Change
NH2+O<=>HNO+H	6.60E+14	-0.5	0	P. Glarborg et al., CNF, 115 (1998) 1 More Recent Data, was Too High
NH2+OH<=>NH+H2O	4.00E+06	2	1001.43	No Change
NH2+NO<=>N2+H2O	2.80E+20	2.654	1258	J.A. Miller et al., IJCK, 1999, 31, 757. More Recent Data, was Too Low
NH2+NO<=>N2H+OH	3.10E+13	-0.48	1180	P. Glarborg et al., CNF, 115 (1998) 1 More Recent Data, was Too High
NH3+M<=>NH2+H+M Unity Efficiencies	2.20E+16	0	93451.24	No Change
NH3+H<=>NH2+H2	6.40E+05	2.39	10181.64	No Change
NH3+O<=>NH2+OH	9.40E+06	1.94	6465.11	No Change
NH3+OH<=>NH2+H2O	2.04E+06	2.04	566.44	No Change
N2H(+M)=N2+H(+M) (Lindemann Falloff) Unity Efficiencies	HPL 6.50E+07 LPL 5.00E+13	0	0	J.A. Miller et al., IJCK, 1999, 31, 757. Previously LPL Only and Too High
N2H+O<=>N2O+H	1.00E+14	0	0	No Change
N2H+OH<=>N2+H2O	5.00E+13	0	0	No Change
H+NO+M<=>HNO+M (Lindemann Falloff) N2/2.00/ O2/2.00/ H2/2.00/ H2O/10.00/ HNO+H<=>NO+H2	HPL 1.50E+15 LPL 2.30E+14	-0.4	0.00E+00	P.Dagaut et al. PECS, 34 (2008) 1-46 Previously Reverse, LPL Only, and Too Low
HNO+OH<=>NO+H2O	4.40E+11	0.72	650.1	No Change
HNO+OH<=>NO+H2O	3.60E+13	0	0	No Change
N2O(+M)<=>N2+O(+M) (Lindemann Falloff) Unity Efficiencies	HPL 8.00E+11 LPL 2E+14	0	62619.5	No Change
N2O+H<=>N2+OH DUP	3.31E+10 7.83E+14	0	5.09E+03	D. L. Baulch et al., JPC, 34 (2005) 757-1397 Previously Not Duplicated and Too Low
N2O+O<=>2 NO	9.15E+13	0	27693	Meagher et al. JPC, A 2000, 104, 6013 More Recent Data, was Too High
N2O+OH<=>N2+HO2	2.00E+12	0	40000	Mueller et al. IJCK, 32 (2000) 317 More Recent Data, was Too High
NO2+M<=>NO+O+M H2O/16.25/	1.10E+16	0	65965.58	No Change
NO+HO2<=>NO2+OH	2.10E+12	0	-480.4	No Change
NO2+H<=>NO+OH	3.50E+14	0	1500.96	No Change
NO2+O<=>NO+O2	1.00E+13	0	599.9	No Change
<b>2 New Steps (Sensitivity Coefficients &gt; 10^-2) Added to Improve Flame Speed Calculation</b>				
NH2+N<=>N2+H+H	7.00E+13	0	0	Glarborg P et al., CNF, 1998;115:1
N2H+O2<=>N2+HO2	2.00E+14	0	0	Glarborg P et al., CNF, 1998;115:1
<b>4 New Steps Added to Improve the Ignition Delay Time Calculation</b>				
NH2+O2<=>HNO+OH	6.20E+07	1.20E+00	35099.9044	Dean A et al. GCC, 2000; p. 125-341.
NH2+O2<=>H2NO+O	2.50E+11	4.80E-01	29586	Dean A et al. GCC, 2000; p. 125-341.
H2NO+O<=>HNO+OH	3.00E+07	2.00E+00	2000	Glarborg P et al., CNF, 1998;115:1
H2NO+O2<=>HNO+HO2	3.00E+12	0.00E+00	25000	J.A. Miller et al., IJCK, 1999, 31, 757.